

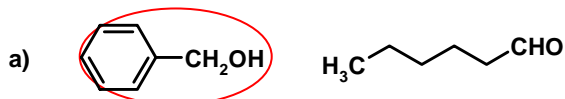
**DEPARTMENT OF MEDICINAL CHEMISTRY  
SCHOOL OF PHARMACY**

Medicinal Chemistry I  
Dr. Umesh R. Desai

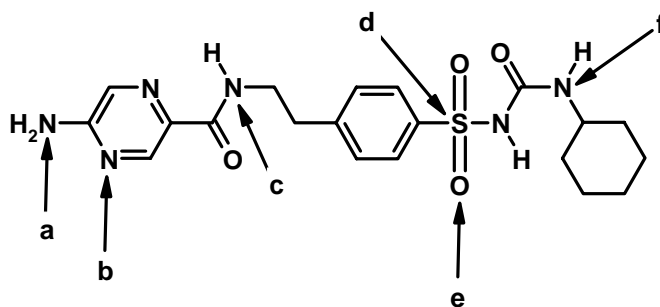
MEDC 501  
September 13, 2006

<b>KEY</b>		<b>IV</b>	<b>V</b>	<b>VI</b>	<b>VII</b>
	<b>2</b>	C	N	O	F
	<b>3</b>	Si	P	S	Cl
	<b>4</b>	Ge	As	Se	Br
	<b>5</b>	Sn	Sb	Te	I
<b>STUDENT NAME</b>		<b>HONOR PLEDGE</b>			

1. Circle the compound with higher boiling/melting point in the following pairs. (6 pts)



2. Glipizide (below) is an oral hypoglycemic agent. Write the Kier – Hall electronegativity value and hybridization state of each non-hydrogen atom marked 'a' through 'f' (6 pts)



Kier-Hall electronegativity	Hybridization State
a = <u>2/4</u>	<u>sp3</u>
b = <u>3/4</u>	<u>sp2</u>
c = <u>2/4</u>	<u>sp3</u>
d = <u>4/9</u>	<u>do not answer</u>
e = <u>5/4</u>	<u>sp2</u>
f = <u>2/4</u>	<u>sp3</u>

3. An inductive effect is the transfer of electronegative negative effect of an atom or a group of atoms to the neighboring bond through sigma bonds resulting in more or less electron withdrawal.

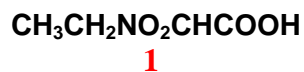
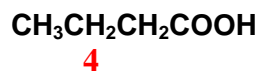
\_\_\_\_\_. (Define in one sentence) (4 pts)

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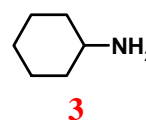
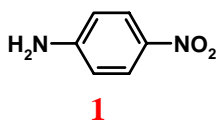
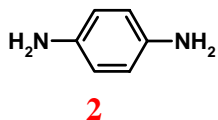
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4. Rank these molecules according to their pKa values. (1 for least pKa value and 4 for highest) (8 pts)

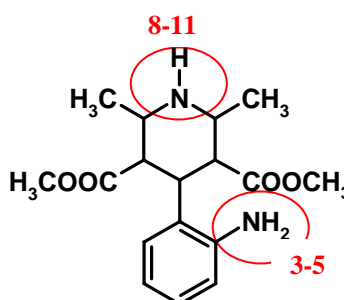
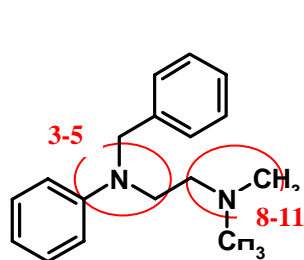
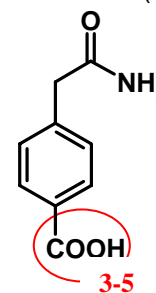
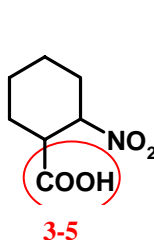
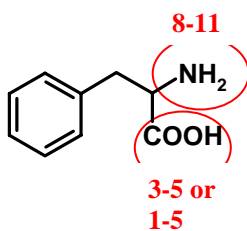


5. Two examples of electron donating groups (for resonance effect), which are electron withdrawing groups for inductive effect are \_\_\_\_\_ and \_\_\_\_\_ (4 pts)  
**-OH, -OCH<sub>3</sub>, -NH<sub>2</sub>, -NHCH<sub>3</sub>, -Cl, -Br, -F, -I**

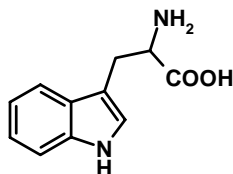
6. Rank the following molecules according to their pK<sub>A</sub> values for the -NH<sub>2</sub> group (1 for the least pKa value and 3 for the highest) (6 pts)



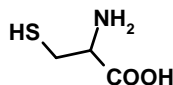
7. In the following structures, circle an ionizable functional group(s) (pH range 0 – 14) and indicate their approximate pK<sub>A</sub> value. **Please NOTE. -1 point for every wrong answer!** (8 pts)



8. Identify the common name of the following natural amino acid residues. (8 pts)



**Tryptophan**



**Cysteine**