

**DEPARTMENT OF MEDICINAL CHEMISTRY  
SCHOOL OF PHARMACY**

Medicinal Chemistry I  
Dr. Umesh R. Desai

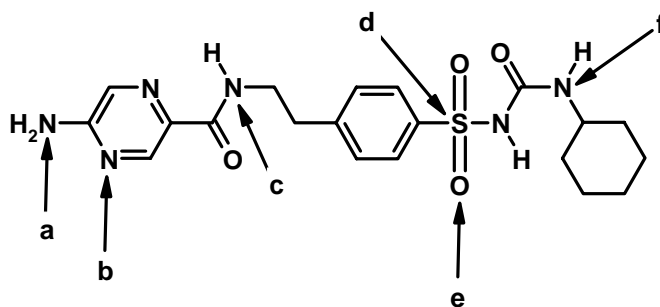
MEDC 501  
September 13, 2006

<b>STUDENT NAME</b> _____	<b>HONOR PLEDGE</b> _____		<b>IV</b>	<b>V</b>	<b>VI</b>	<b>VII</b>
		<b>2</b>	C	N	O	F
		<b>3</b>	Si	P	S	Cl
		<b>4</b>	Ge	As	Se	Br
		<b>5</b>	Sn	Sb	Te	I

1. Circle the compound with higher boiling/melting point in the following pairs. (6 pts)



2. Glipizide (below) is an oral hypoglycemic agent. Write the Kier – Hall electronegativity value and hybridization state of each non-hydrogen atom marked 'a' through 'f' (6 pts)



Kier-Hall electronegativity	Hybridization State
<b>a</b> = _____	_____
<b>b</b> = _____	_____
<b>c</b> = _____	_____
<b>d</b> = _____	do not answer
<b>e</b> = _____	_____
<b>f</b> = _____	_____

3. An inductive effect is

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

(Define in one sentence)

(4 pts)

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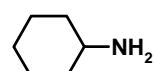
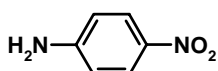
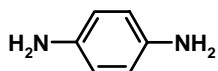
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4. Rank these molecules according to their pKa values. (1 for least pKa value and 4 for highest) (8 pts)

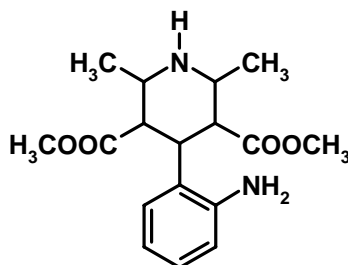
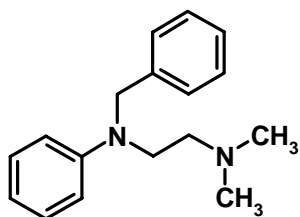
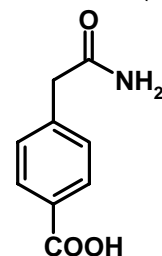
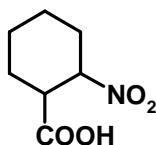
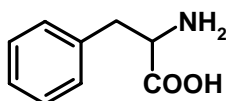


5. Two examples of electron donating groups (for resonance effect), which are electron withdrawing groups for inductive effect are \_\_\_\_\_ and \_\_\_\_\_ (4 pts)

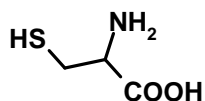
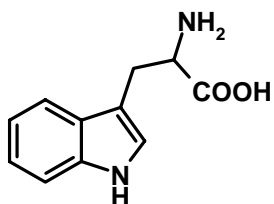
6. Rank the following molecules according to their pKa values for the  $-\text{NH}_2$  group (1 for the least pKa value and 3 for the highest) (6 pts)



7. In the following structures, circle an ionizable functional group(s) (pH range 0 – 14) and indicate their approximate pKa value. **Please NOTE. -1 point for every wrong answer!** (8 pts)



8. Identify the common name of the following natural amino acid residues. (8 pts)



\_\_\_\_\_

\_\_\_\_\_