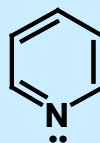


Heterocycles

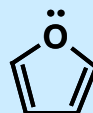
Hückel's Rule for Aromaticity

- $4n+2$ π -electrons
- planarity of the structure

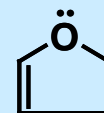
Aromatic?



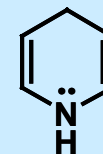
Y



Y

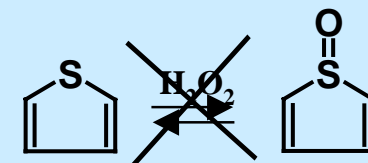
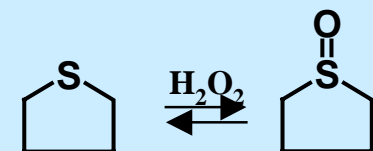
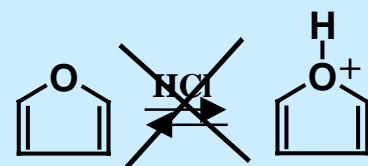
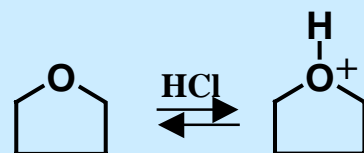
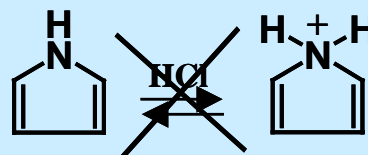


N



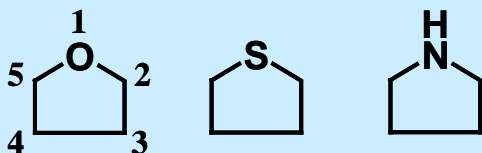
N

- ✓ Difference between reactions of aromatic and non-aromatic heterocycles

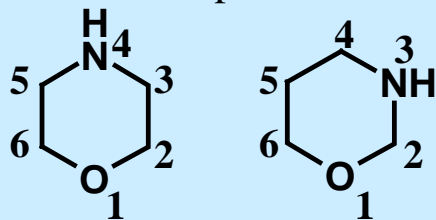


Nomenclature of Heterocyclic Compounds

1. Heteroatom is to be counted as 1 or as low as possible



2. When there is more than one heteroatom, preference is given to O, then S, then N, then C. Also N-H presides over $\text{N}=\text{}$. [Mnemonic: Old Soldiers Never Cry]

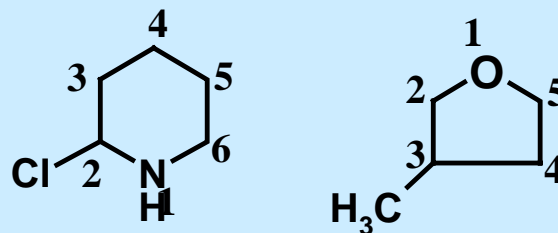


3. When there is more than one heteroatom, numbering should be as direct as possible from one to the other



Nomenclature of Heterocyclic Compounds

4. Substituents are numbered as low as possible

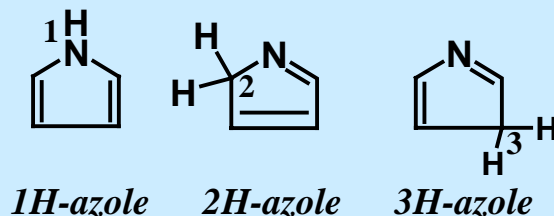


5. Acceptable prefixes for common heteroatoms: **O** = Oxa; **N** = Aza; **S** = Thia
 6. Common suffixes for N- and non-N-heterocycles

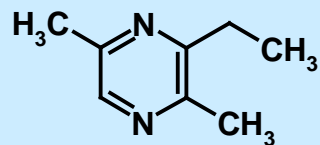
<i>Ring Size</i>	<i>Saturated</i>	<i>Partly Saturated</i>	<i>Unsaturated</i>
<i>Rings with N</i>			
3	-iridine		-irine
4	-etidine		-ete
5	-olidine	-oline	-ole
6	-ine	(di or tetrahydro)	-ine
7	(hexahydro)	(di or tetrahydro)	-epine
8	(octahydro)	(di, tetra, or hexahydro)	-ocine
<i>Rings without N</i>			
3	-irane		-irene
4	-etane		-ete
5	-olane	-olene	-ole
6	-ane	(di or tetrahydro)	-ine
7	-epane	(di or tetrahydro)	-epine
8	-ocane	(di, tetra, or hexahydro)	-ocin

Nomenclature of Heterocyclic Compounds

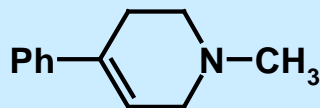
7. For partially unsaturated systems, H is(are) are used to indicate the location of saturation



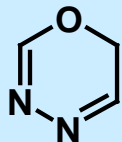
Examples:



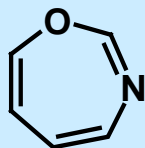
Common Name: 2-ethyl-3,6-dimethyl-pyrazine
IUPAC: 2-ethyl-3,6-dimethyl-1,4-diazine



Common Name: 1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine



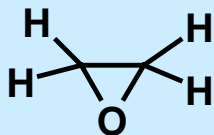
Common Name: 6H-diazapyran
IUPAC: 6H-1,3,4-oxadiazine



Common Name: 1,3-oxazepine

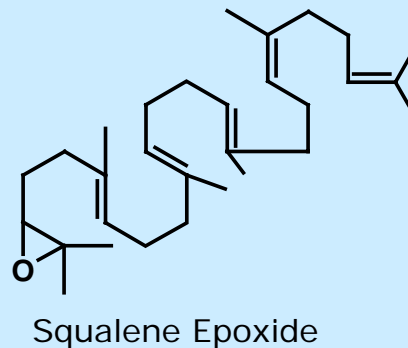
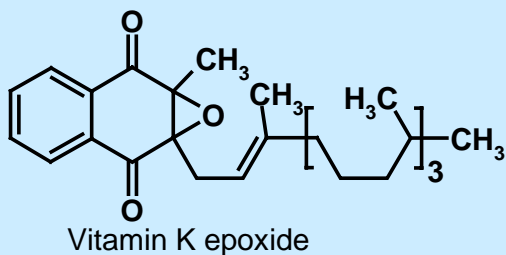
Oxygen Containing Heterocycles

✓ 3-Membered Ring :: Ethylene Oxide or oxirane



- Functional group is epoxide
- planar structure
- gas at room temperature
- used as sterilant for operating room equipment
- extremely toxic to living cell and is thought to be mutagen and probably a carcinogen

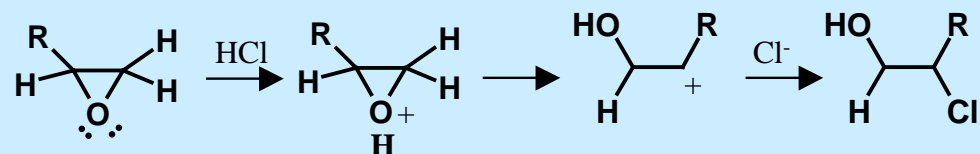
✓ Examples of epoxides in our System



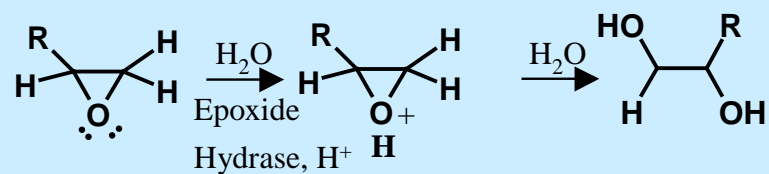
Oxygen Containing Heterocycles

✓ Reactions of Epoxides

Acid Opening of Epoxides



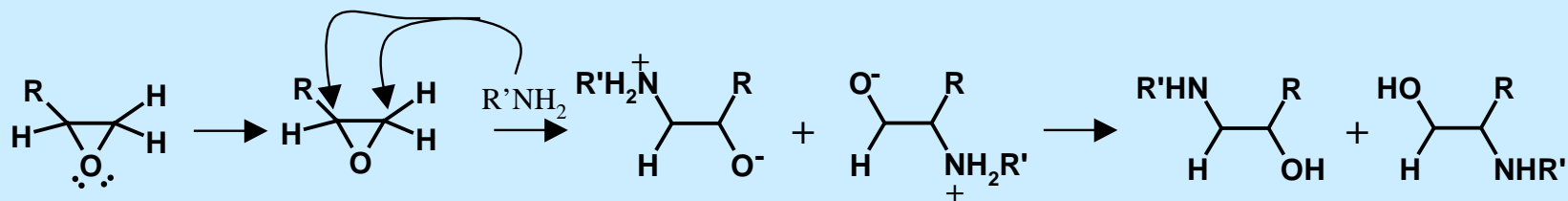
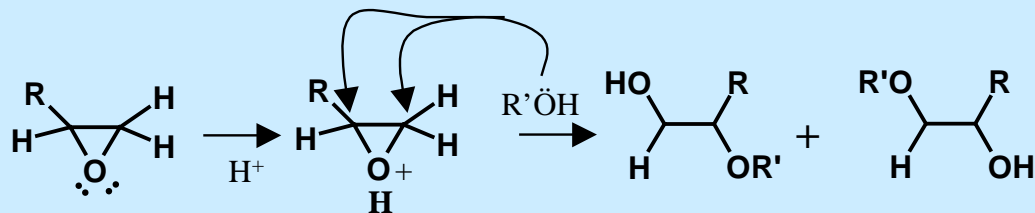
Enzymatic Opening of Epoxides



Oxygen Containing Heterocycles

✓ Reactions of Epoxides

Opening of Epoxides with Amines and Alcohols



Oxygen Containing Heterocycles

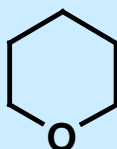
✓ Higher Oxides



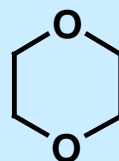
Oxetane



oxolane



oxane

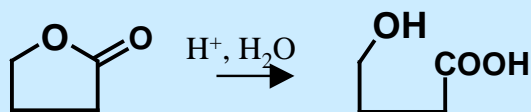


1,4-dioxane

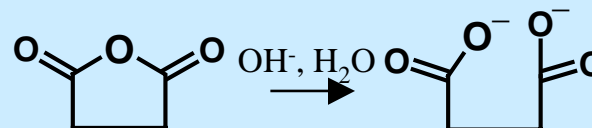
Properties:

- *all non-planar*
- *oxetane has a small pucker*
- *all good solvents and used in place of diethylether*
- *all are generally non-toxic, except for dioxane, which is mutagenic*

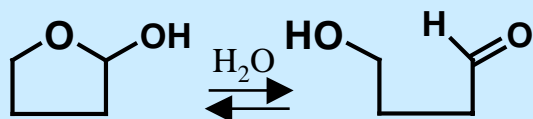
Stability of these oxides change drastically on α -substitution



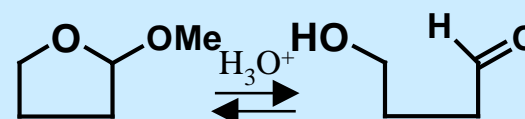
Lactones



anhydrides



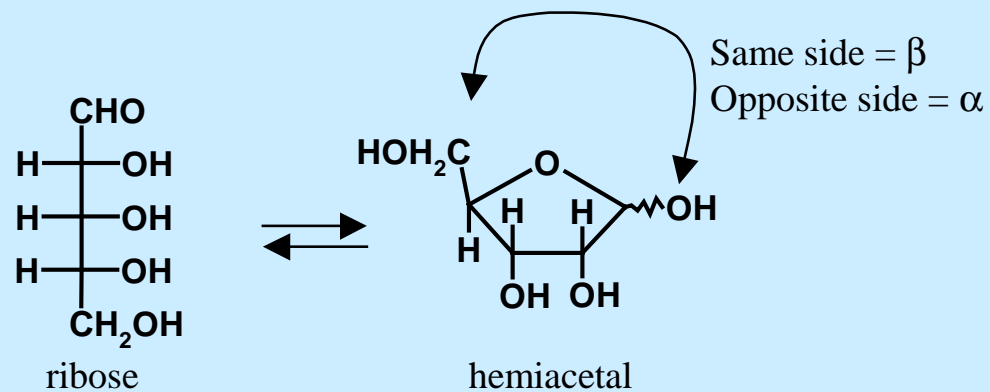
Hemi-acetals



acetals

Oxygen Containing Heterocycles

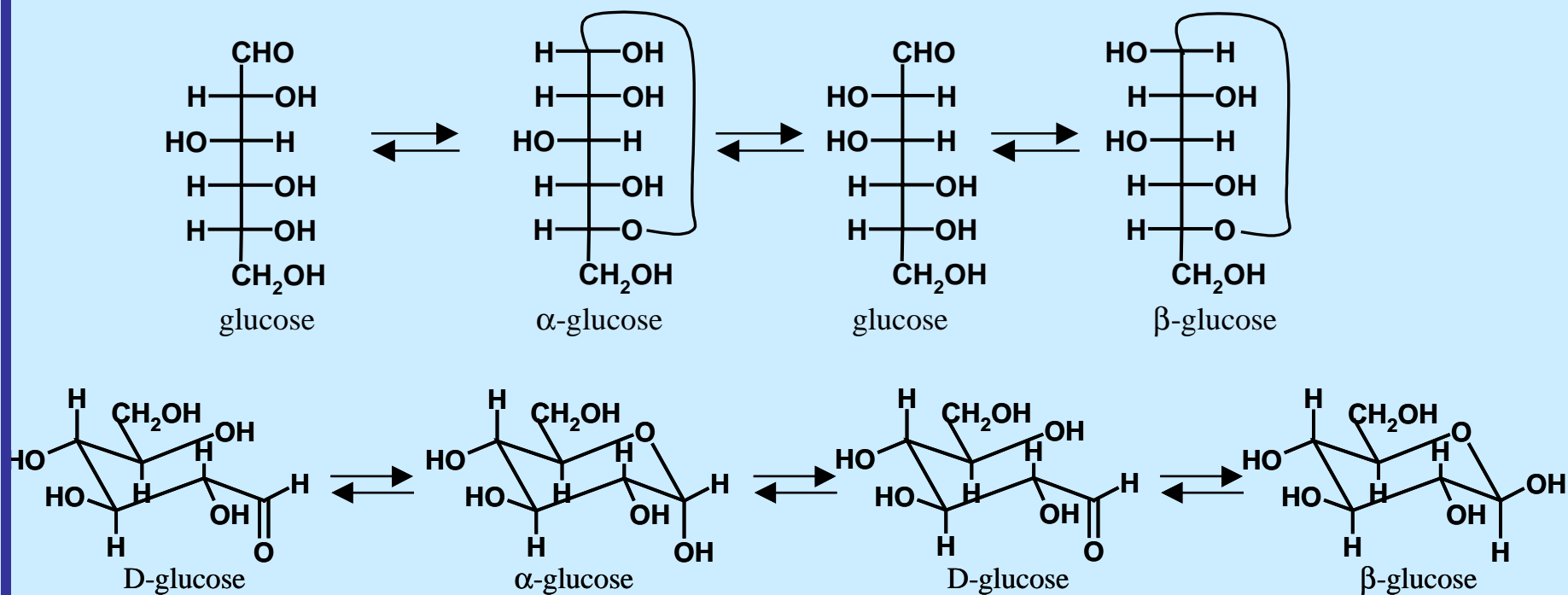
✓ Examples of Acetals or Hemi-Acetals in Nature



Oxygen Containing Heterocycles

✓ Interconversion of α - and β -forms of sugars

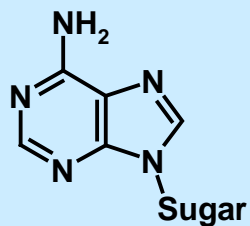
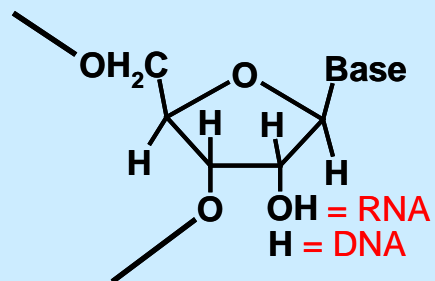
The phenomenon of mutarotation



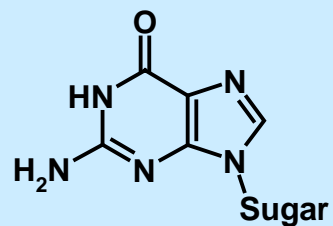
Oxygen Containing Heterocycles

✓ Examples of sugars in natural systems

Deoxynucleic Acids and Ribonucleic Acids



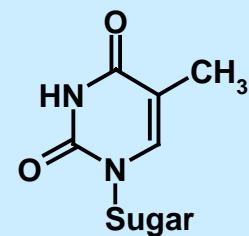
Adenine



Guanine



Cytidine

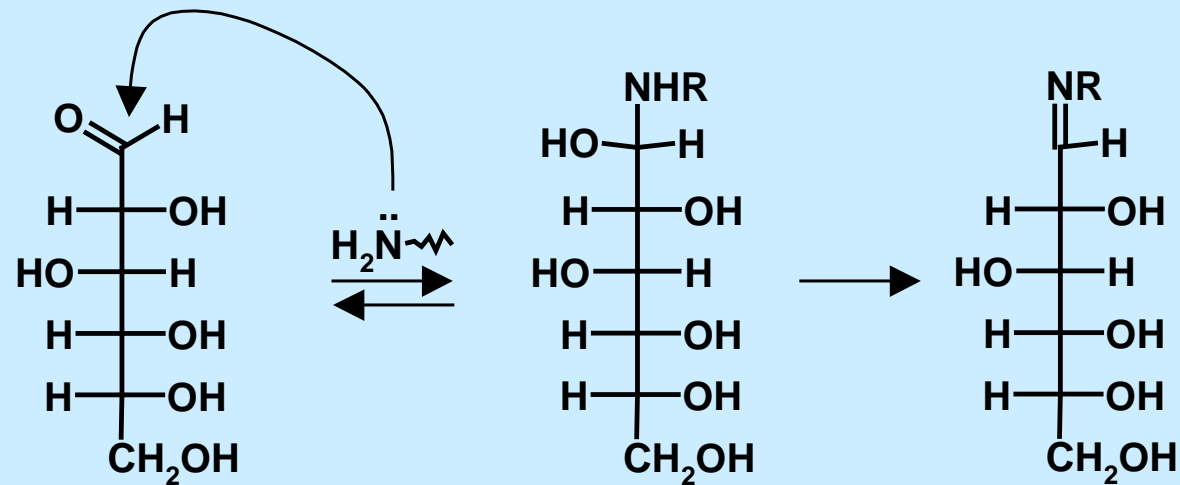


Thymidine

Live link: <http://www.bmrb.wisc.edu/referenc/nomenclature/>

Oxygen Containing Heterocycles

✓ Sugars and Diabetes

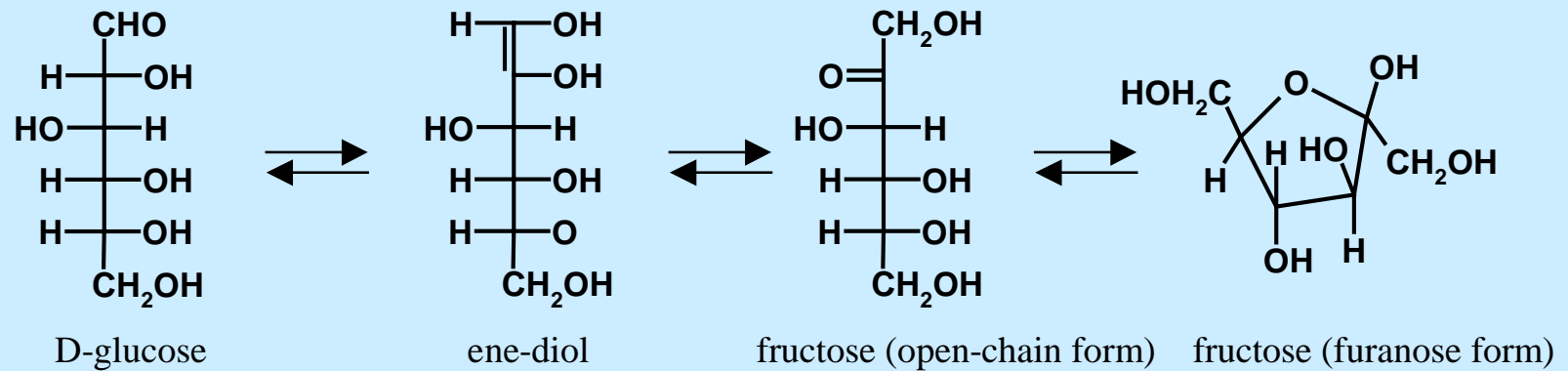


Glycated Hemoglobin (HbA1c or GHb)

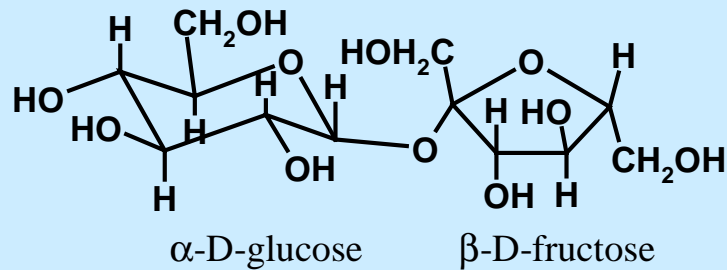
Also called glycosylated hemoglobin. The saccharide equilibrium is the reason for the formation of glycated hemoglobin in diabetic patients. Excess free plasma sugar, and therefore the open-chain aldehyde, reacts with the N-terminal NH_2 group of hemoglobin (or lysine side chains) to form glycated adducts, which are detected to confirm and monitor diabetes. The upper limit for a normal individual is 7% HbA1c.

Oxygen Containing Heterocycles

✓ Fructose

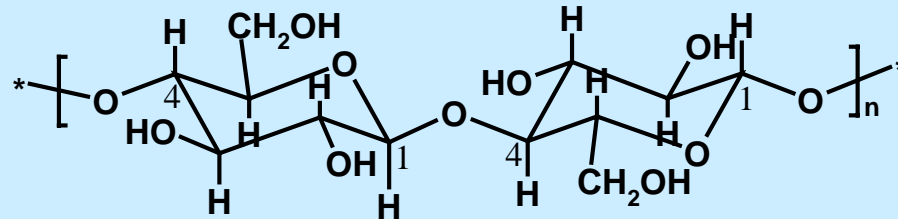


✓ Sucrose

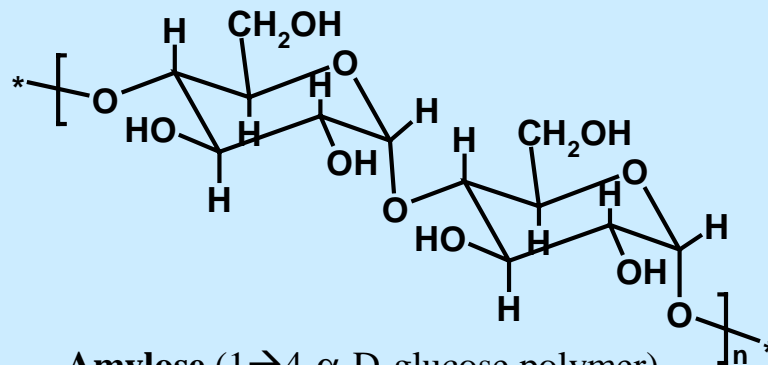


Oxygen Containing Heterocycles

✓ Polysaccharides – Cellulose and Starch



Cellulose (1→4-β-D-glucose polymer)



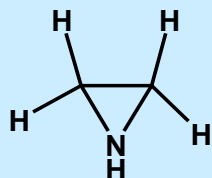
Amylose (1→4-α-D-glucose polymer)

80% constituent of starch; 20% is amylopectin

Nitrogen Containing Heterocycles

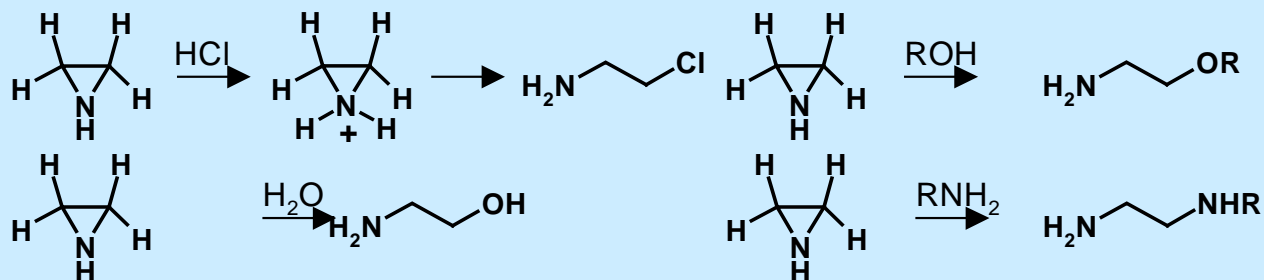
✓ Three-membered

Aziridine



Planar
Highly strained; highly reactive
Toxic and carcinogenic

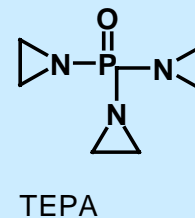
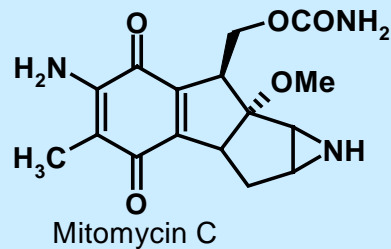
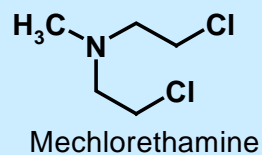
Reactions of Aziridine



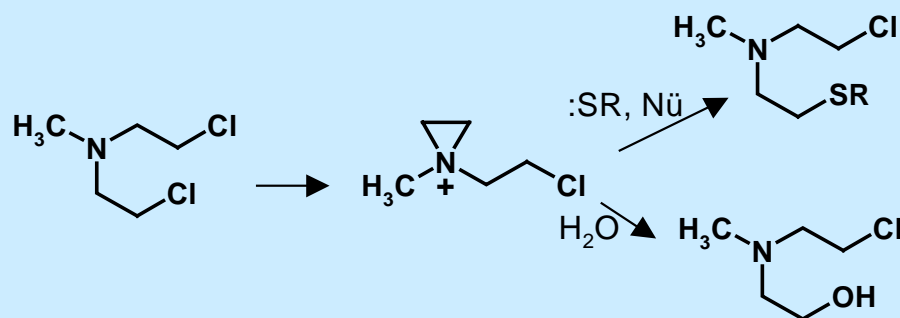
Nitrogen Containing Heterocycles

✓ Three-membered

Examples of Aziridine Containing Drugs



Mechanism of Action of Mechlorethamine – Formation of an Aziridinium Ion

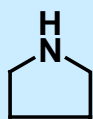


Nitrogen Containing Heterocycles

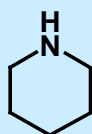
✓ Higher Membered Ring Systems - Saturated



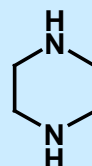
azetidine



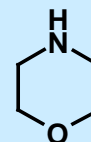
pyrrolidine
(azolidine)



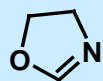
piperidine
(perhydroazine)



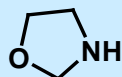
piperazine
(perhydro-1,4-diazine)



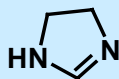
morpholine
(perhydro-1,4-oxazine)



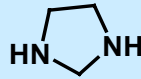
Oxazoline



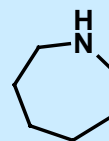
Oxazolidine



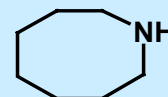
Imidazoline



Imidazolidine



Hexahydroazepine



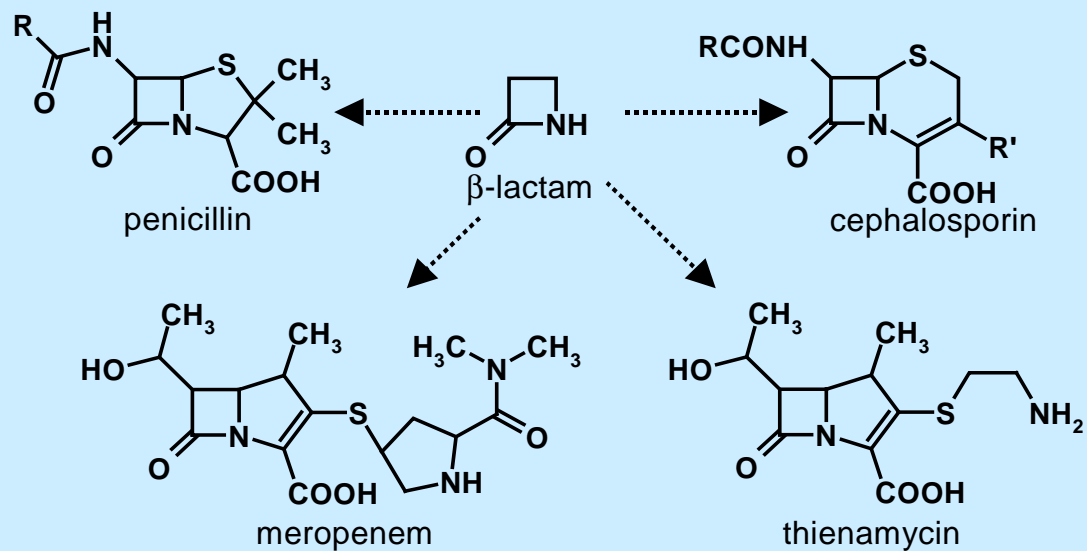
Octahydroazepine

- ✓ *Non-planar molecules*
- ✓ *pK_A same as secondary amines (~8-11)*
- ✓ *Most are liquids with fish-like odor*
- ✓ *Salts are odorless and crystalline*
- ✓ *Some have two pK_A values, e.g., first $(pK_A)_1 > (pK_A)_2$ for imidazolidine*

Nitrogen Containing Heterocycles

✓ Higher Membered Ring Systems - Saturated

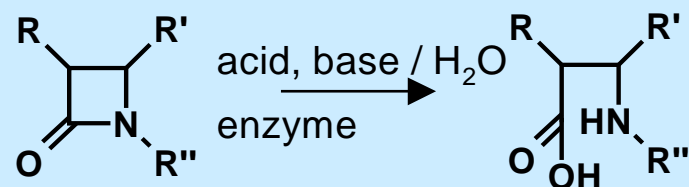
Examples of 4-membered nitrogen containing rings in drugs



Nitrogen Containing Heterocycles

✓ Higher Membered Ring Systems - Saturated

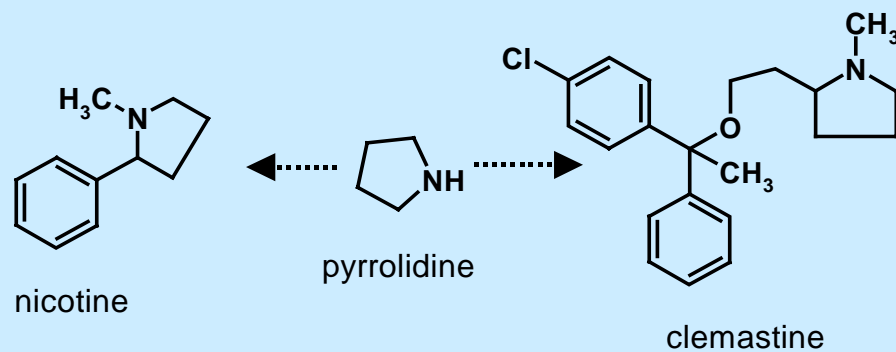
Hydrolysis of β -lactams



Nitrogen Containing Heterocycles

✓ Higher-Membered Ring Systems - Saturated

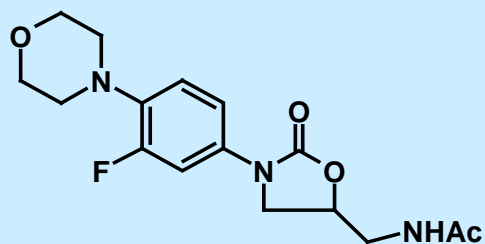
Examples of 5-membered nitrogen containing ring in drugs



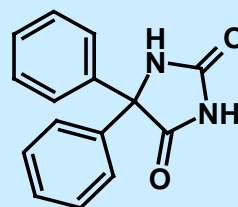
Nitrogen Containing Heterocycles

✓ Higher-Membered Ring Systems - Saturated

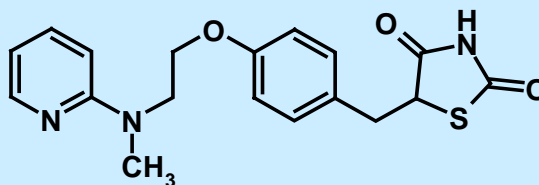
Examples of 5-membered ring containing two heteroatoms in drugs



linezolid
(Anti-infection)



phenytoin
(anti-psychotic)

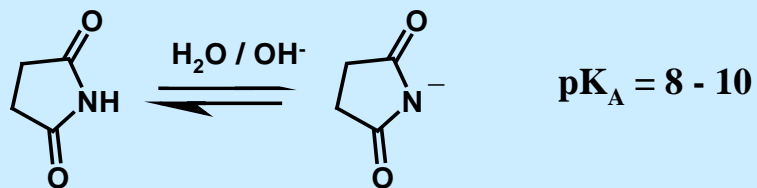


Rosiglitazone
(pyridine + thiazolidine2,4-dione)
Anti-diabetic

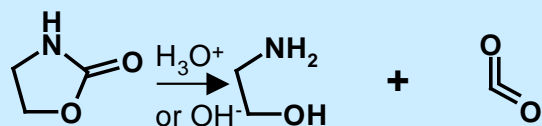
Nitrogen Containing Heterocycles

✓ Higher-Membered Ring Systems - Saturated

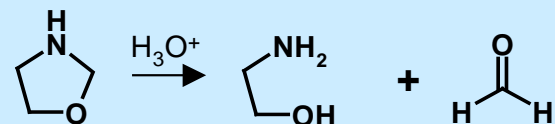
Acidity of -NH in Imides



Instability of Carbamates



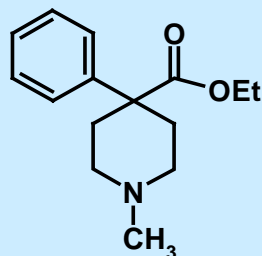
Instability of Oxazolidines



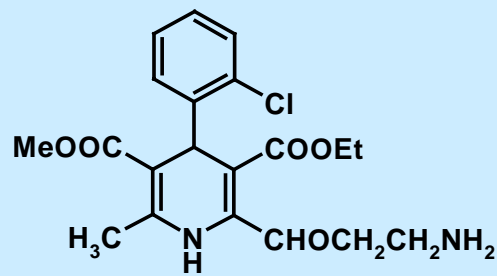
Nitrogen Containing Heterocycles

✓ Higher-Membered Ring Systems - Saturated

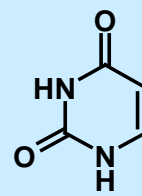
6-membered systems – piperidines, dihydropyridines



Meperidine
(piperidine)
relieves pain,
narcotic analgesic



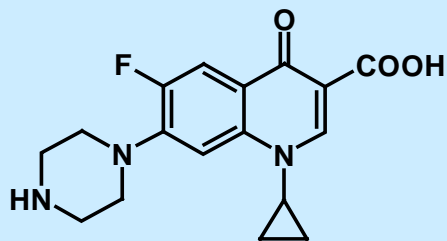
Amlodipine
(dihydropyridine)
 Ca^{2+} channel blocker,
Antihypertensive, antianginal



Uracil
(pyrimidine)
One of the bases
of RNA and DNA



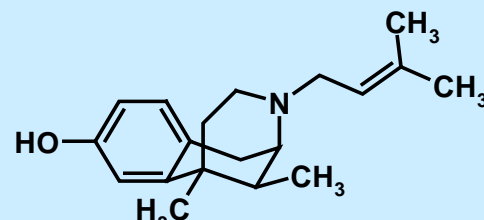
Phenobarbital
(pyrimidine)
controls epilepsy, seizures,
as a sedative to relieve anxiety



Ciprofloxacin
(piperazine)
Antibacterial
(anthrax)



Phenmetrazine
(morpholine)
Appetite suppressant
CNS stimulant
Amphetamine-like

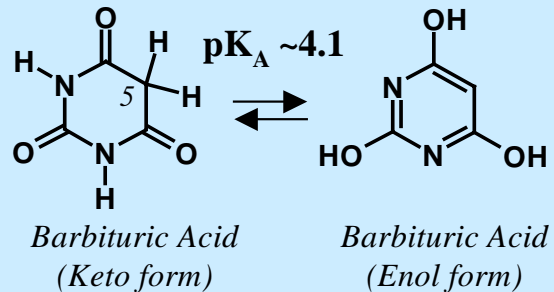
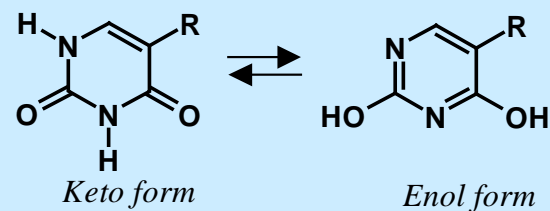
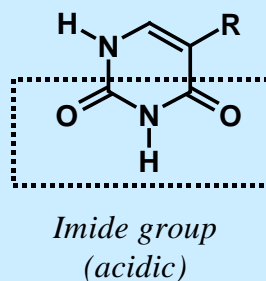
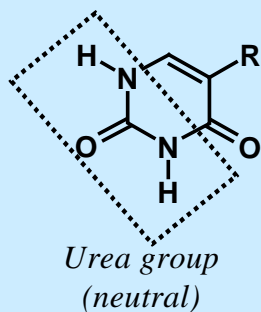


Pentazocine
(octahydroazepine)
Pain reliever

Nitrogen Containing Heterocycles

✓ Higher-Membered Ring Systems - Saturated

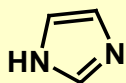
Tautomerism in Barbituric Acids – Enhanced Acidity of -NH



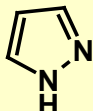
Nitrogen Containing Heterocycles

✓ Unsaturated Ring Systems

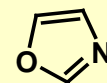
5-membered ring systems



Imidazole
(1,3-diazole)



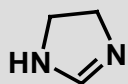
Pyrazole
(1,2-diazole)



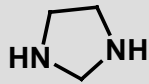
Oxazole
(1,3-oxazole)



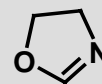
Isoxazole
(1,2-oxazole)



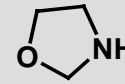
Imidazoline
(1,3-diazoline)



Imidazolidine
(1,3-diazolidine)



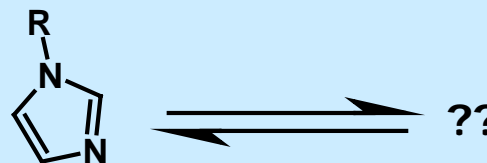
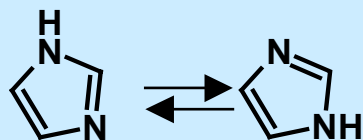
Oxazoline
(1,3-oxazoline)



Oxazolidine
(1,3-oxazolidine)

*Partly/fully saturated ring
structures here are only for reference*

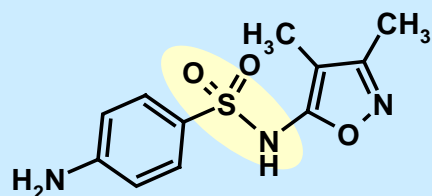
Resonance in imidazole – the movement of a hydrogen



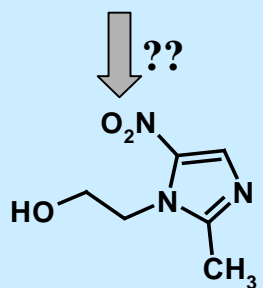
Nitrogen Containing Heterocycles

✓ Unsaturated Ring Systems

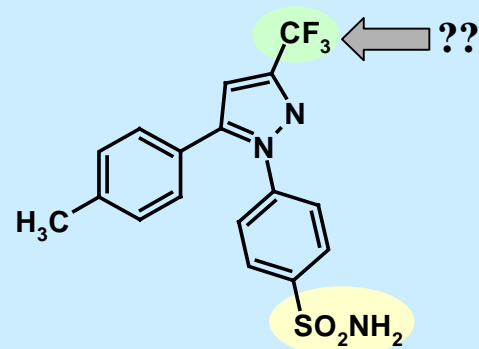
Examples of these ring systems in drugs



Sulfisoxazole
sulfa-drug
anti-bacterial



Metronidazole
anti-bacterial,
anti-protozoan



Celecoxib
COX-2 inhibitor
anti-inflammatory

Heterocycles

✓ Unsaturated Ring Systems Containing Sulfur

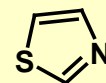
5-membered ring systems



Tetrahydrothiophene
(thiolane)



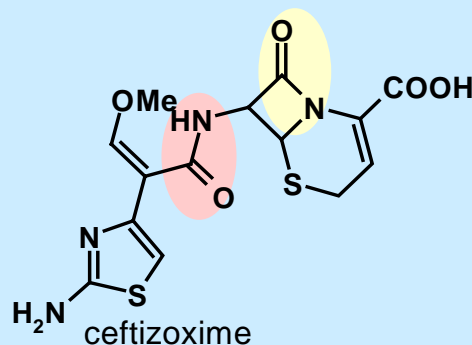
Thiophene
(Thiole)



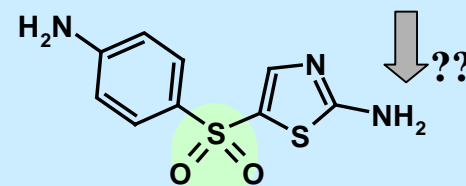
1,3-thiazole



Methapyrilene
Anti-histaminic



ceftizoxime
Anti-biotic

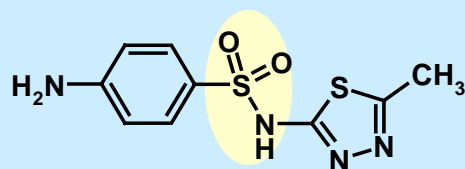


thiazolsulfone
anti-malarial

Heterocycles

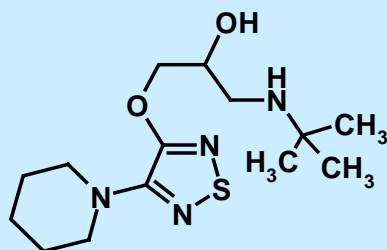
✓ More Complex Unsaturated Ring Systems

- the 1,3,4- and 1,2,5-thiadiazoles
- the 1,3,4-triazoles
- the 1,2,3,5-tetrazoles



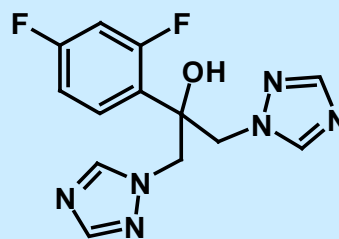
sulfamethizole

Anti-bacterial
Anti-protozoan



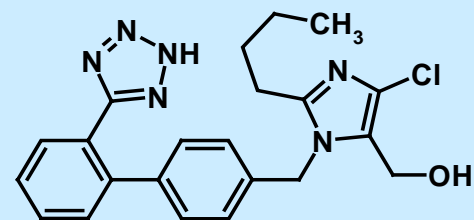
timolol

Anti-hypertensive



fuconazole

Anti-fungal



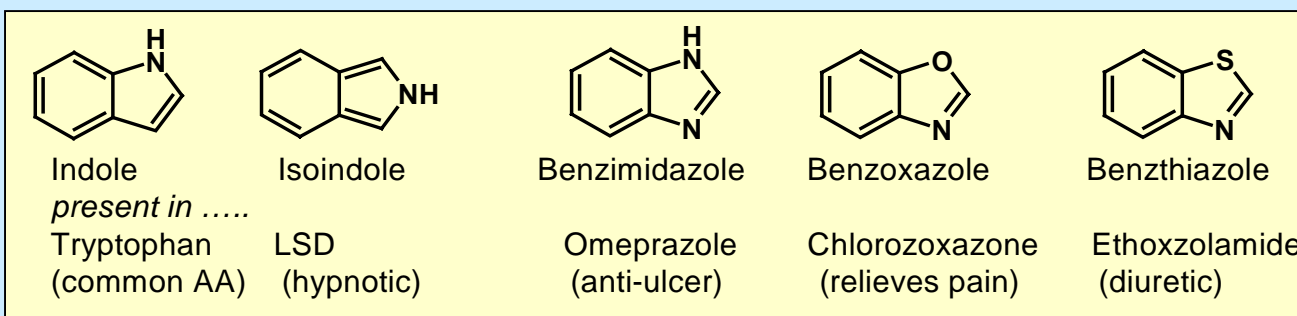
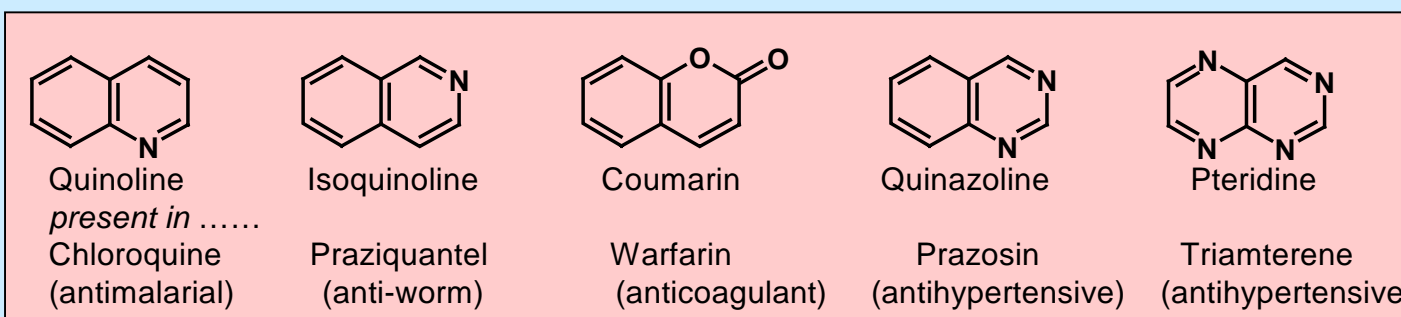
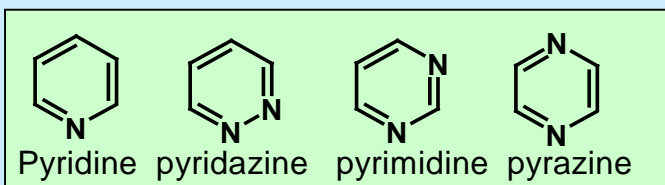
losartan

Anti-hypertensive

Heterocycles

✓ More Complex Unsaturated Ring Systems

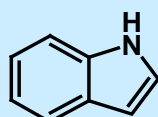
➤ 6-membered and higher heterocycles



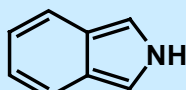
Heterocycles

✓ More Complex Unsaturated Ring Systems

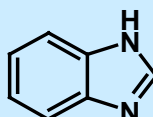
➤ 6-membered and higher heterocycles



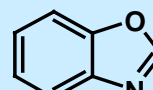
Indole
present in
Tryptophan
(common AA)



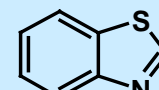
Isoindole
LSD
(hypnotic)



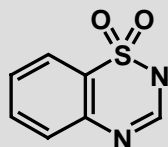
Benzimidazole
Omeprazole
(anti-ulcer)



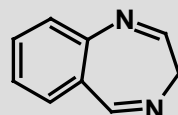
Benzoxazole
Chlorzoxazone
(relieves pain)



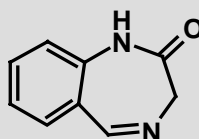
Benzthiazole
Ethoxzolamide
(diuretic)



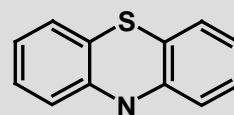
1,2,4-benzothia
diazin-1,1,-dioxide
present in
Chlorothiazide
(antihypertensive)



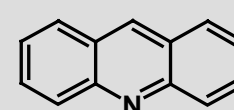
3H-1,4-Benzo
diazepine
Chlordiazepoxide
(sedative, hypnotic)



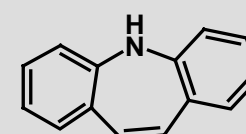
1,3-Dihydro-2H-1,4-
benzodiazepin-2-one
Diazepam
(sedative, anxiolytic)



Phenothiazine
Chlorpromazine
(anxiolytic)



Acridine
Quinocrine
(antibiotic)



5H-dibenz
[b,f]azepine
Imipramine
(anti-depressants)

Heterocycles

✓ Electrophilic Aromatic Substitution in Heterocycles

- heterocycles generally react well with electrophiles; better than benzene (except for pyridines)
- their stability to metabolic enzymes is generally lower than their carbocyclic analogs
- EAS occurs at the 2-position of 5-membered heterocycles
- EAS occurs at the 3-position of 6-membered heterocycles

